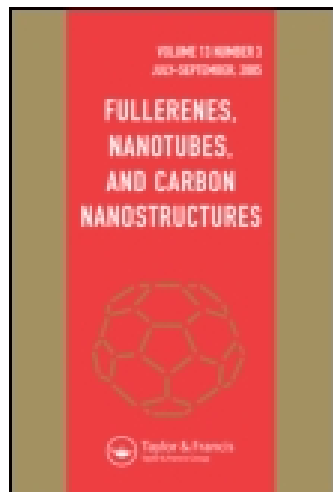


This article was downloaded by: [University of Victoria]

On: 29 September 2014, At: 17:15

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Fullerenes, Nanotubes and Carbon Nanostructures

Publication details, including instructions for authors and subscription information:

<http://www.tandfonline.com/loi/lfn20>

### Mass Spectrometric Transmutation of Fullerenes

William Henderson<sup>a</sup> & J. Scott McIndoe<sup>b</sup>

<sup>a</sup> Department of Chemistry, The University of Waikato, Hamilton, New Zealand

<sup>b</sup> Department of Chemistry, The University of Victoria, Victoria, BC, Canada

Accepted author version posted online: 02 Dec 2013. Published online: 05 Mar 2014.

To cite this article: William Henderson & J. Scott McIndoe (2014) Mass Spectrometric Transmutation of Fullerenes, Fullerenes, Nanotubes and Carbon Nanostructures, 22:7, 663-669, DOI: [10.1080/1536383X.2012.717558](http://dx.doi.org/10.1080/1536383X.2012.717558)

To link to this article: <http://dx.doi.org/10.1080/1536383X.2012.717558>

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at <http://www.tandfonline.com/page/terms-and-conditions>

# Mass Spectrometric Transmutation of Fullerenes

WILLIAM HENDERSON<sup>1</sup> AND J. SCOTT McINDOE<sup>2</sup>

<sup>1</sup>Department of Chemistry, The University of Waikato, Hamilton, New Zealand

<sup>2</sup>Department of Chemistry, The University of Victoria, Victoria, BC, Canada

*MS/MS studies of the higher fullerenes C<sub>n</sub><sup>+</sup> (n = 60, 70, all even from 76-94) revealed that certain product ions were especially stable, and that these matched closely to the fullerenes that are particularly abundant in synthetic mixtures. All fullerenes fragmented by loss of C<sub>2</sub> molecules.*

**Keywords** Mass spectrometry, laser desorption, fragmentation, fullerenes

## Introduction

In 2005, Cross and Saunders described the high temperature annealing of higher fullerenes (C<sub>70</sub>, and various isomers of C<sub>76</sub>, C<sub>78</sub>, C<sub>84</sub>) to form smaller fullerenes through loss of carbon atoms (1). C<sub>60</sub> survived the passage through the ~1000°C oven without decomposition. The authors memorably described this transformation as the equivalent of turning gold into lead, given the price differential between pure samples of the higher versus lower fullerenes. The annealing process occurred in the gas phase in an argon flow, and the fullerenes resided in the oven for about 40 seconds. C<sub>76</sub> decomposed to C<sub>70</sub> and small amounts of C<sub>60</sub>, and larger fullerenes also lost carbon atoms to form smaller fullerenes.

A MALDI-TOF/TOF mass spectrometer offers the opportunity to study this process from a different angle, and the MS/MS capability allows a mixture of fullerenes to be investigated, rather than having to laboriously separate the higher fullerenes. We analysed a commercially available mixture of fullerenes by tandem mass spectrometry without a matrix using a MALDI-TOF/TOF instrument and gathered decomposition information on the fullerenes C<sub>60</sub><sup>+</sup>, C<sub>70</sub><sup>+</sup>, and all C<sub>n</sub><sup>+</sup> (where n = even number) from C<sub>76–94</sub><sup>+</sup>. All appear to decompose by sequential elimination of C<sub>2</sub> molecules, but certain fullerenes consistently dominate the suite of daughter ions. Experiments were run without a matrix, as we have found other inorganic cluster compounds to provide more intense spectra without a matrix (2).

The mass spectrometric study of fullerenes has a long and proud history (3), starting with the Nobel prize-winning discovery of C<sub>60</sub> from mass spectrometric experiments on the laser vaporization of graphite (4). Smalley reported the photodissociation of C<sub>60</sub><sup>+</sup> in TOF/TOF experiments, he showed the primary fragmentation is C<sub>2</sub> loss and that the initial C<sub>2</sub> loss requires activation of 18 eV (5). High-energy collision-induced dissociation studies show a similar distribution of fragments, provided heavier target gases than He are used (e.g. O<sub>2</sub>, N<sub>2</sub>) (6). High-energy (150-1050 eV/molecule) surface impact fragmentation

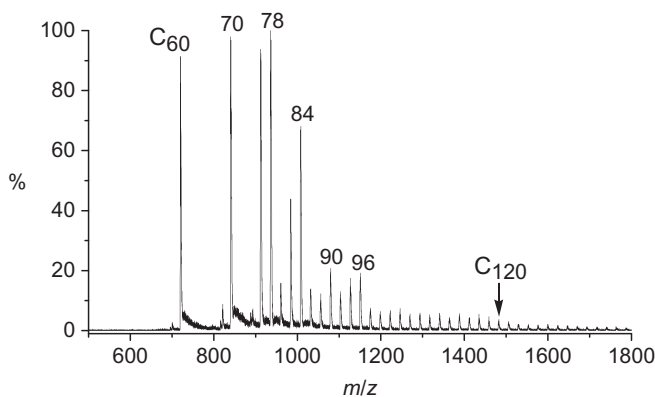
Address correspondence to J. Scott McIndoe, Department of Chemistry, University of Victoria, P.O. Box 3065, Victoria, BC V8W 3V6, Canada. E-mail: mcindoe@uvic.ca; or William Henderson, University of Waikato, Private Bag 3105, Hamilton 3240, New Zealand. E-mail: hende@waikato.ac.nz

of  $C_{60}$ ,  $C_{70}$ ,  $C_{76}$ ,  $C_{84}$ , and  $C_{94}$  on graphite generated even-numbered  $C_n$  species with abundances similar to those observed in fullerene synthesis (7). Cordero, Cornish, and Cotter conducted a MALDI-TOF/TOF investigation of  $C_{60}^+$  and  $C_{70}^+$ , including collisional activation with He, Ar, Xe, and  $SF_6$  (8). High-energy single collisions produced the best-resolved spectra, forming high mass  $C_{2n}^+$  and low mass  $C_n^+$  ions. Our study used solely Ar as the collision gas, and we examined a range of higher fullerenes in addition to  $C_{60}^+$  and  $C_{70}^+$ .

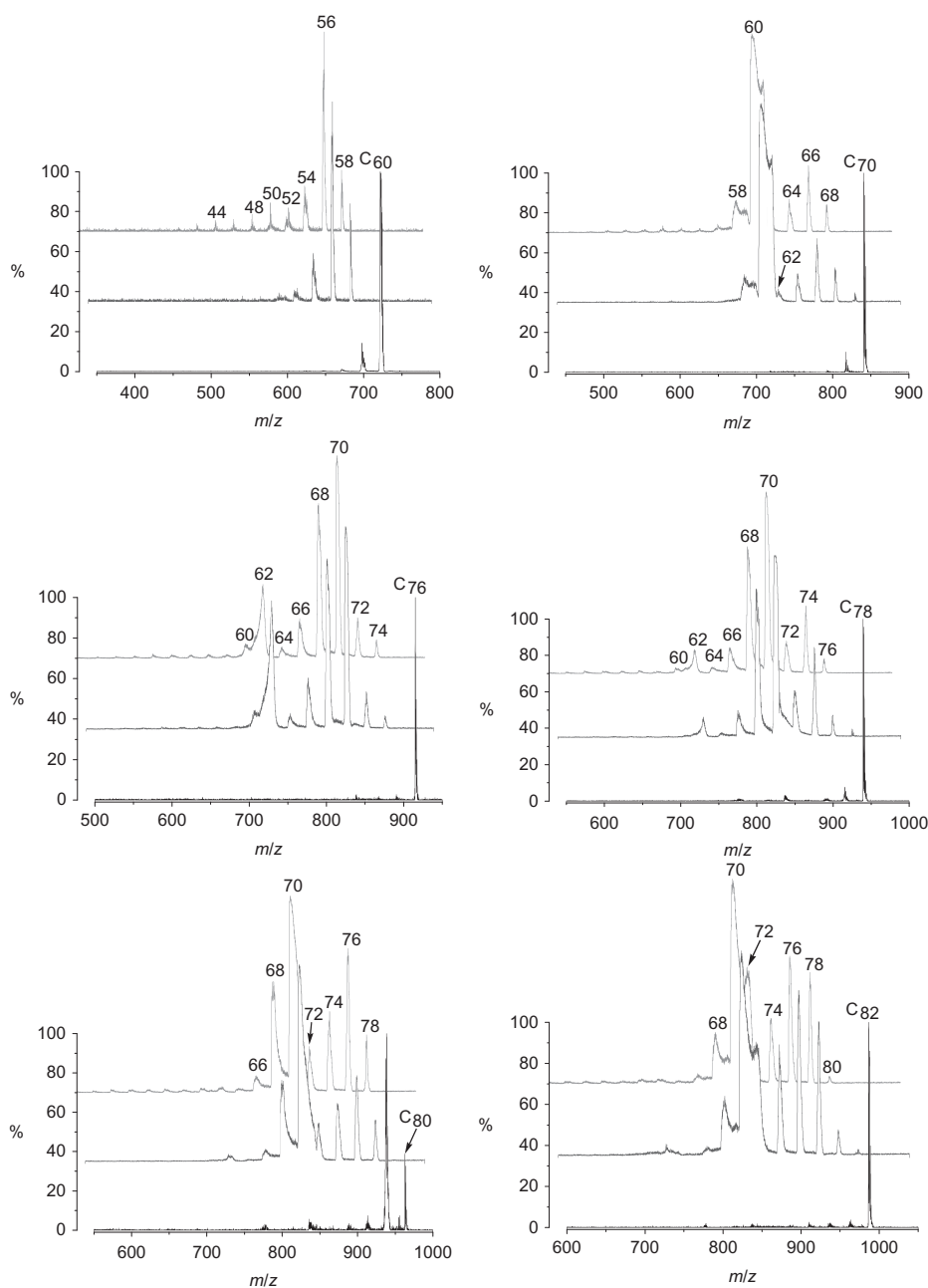
## Results and Discussion

The LDI-TOF mass spectrum of a commercial sample of mixed fullerenes is shown in Figure 1. It consists primarily of  $C_{60}$ ,  $C_{70}$ ,  $C_{76}$ ,  $C_{78}$  and  $C_{84}$  with smaller amounts of higher fullerenes, all with an even number of carbon atoms, up to about  $C_{150}$  at which point the signal-to-noise ratio becomes very low. The intensity of the  $C_n^+$  radical cations was good, with sample preparation involving simply evaporating a  $CS_2$  solution onto an AnchorChip<sup>TM</sup> plate. The spectrum resembles that obtained of crude fullerene soot produced by resistive heating of graphite under an inert atmosphere (9).

Each of these  $C_n^+$  ions may be selected for MS/MS analysis, fragmentation being caused primarily by increasing the intensity of UV laser used for desorption. The excess energy deposited into the ions causes them to fragment *after* acceleration, and fragment ions are separated by the reflectron TOF. Ions decomposing in the first field-free region of the reflectron will appear at an  $m/z$  value different from that of the ion entering the reflectron, and such post-source decay (PSD) peaks appear broadened and at lower  $m/z$  values. Data were collected at three values of fragmentation energy for all fullerenes studied: (a) threshold laser power, (b) high laser power, and (c) high laser power with CID gas (Ar) switched on. The Bruker LIFT experiment was used for all MS/MS studies (10). Precursor ions are selected and fragmented in TOF-1 (the first of two co-linear time-of-flight mass analyzers). Fragment ions are selected to proceed to the source of TOF-2, where they are again accelerated and analyzed. Both sources allow control over acceleration and focusing of ions (11). Sufficient intensity for MS/MS studies were obtained for the fullerenes  $C_{60}$ ,  $C_{70}$ ,  $C_{76}$ ,  $C_{78}$ ,  $C_{80}$ ,  $C_{82}$ ,  $C_{84}$ ,  $C_{86}$ ,  $C_{88}$ ,  $C_{90}$ ,  $C_{92}$ , and  $C_{94}$ . The higher fullerenes provided MS/MS spectra that were too noisy for successful analysis. Mass spectra of  $C_{60}$ ,  $C_{70}$ ,  $C_{76}$ ,  $C_{78}$ ,  $C_{80}$ , and  $C_{82}$  are illustrated in Figure 2, while those for  $C_{84}$ ,  $C_{86}$ ,  $C_{90}$ ,  $C_{92}$ , and  $C_{94}$  are shown in Figure 3.

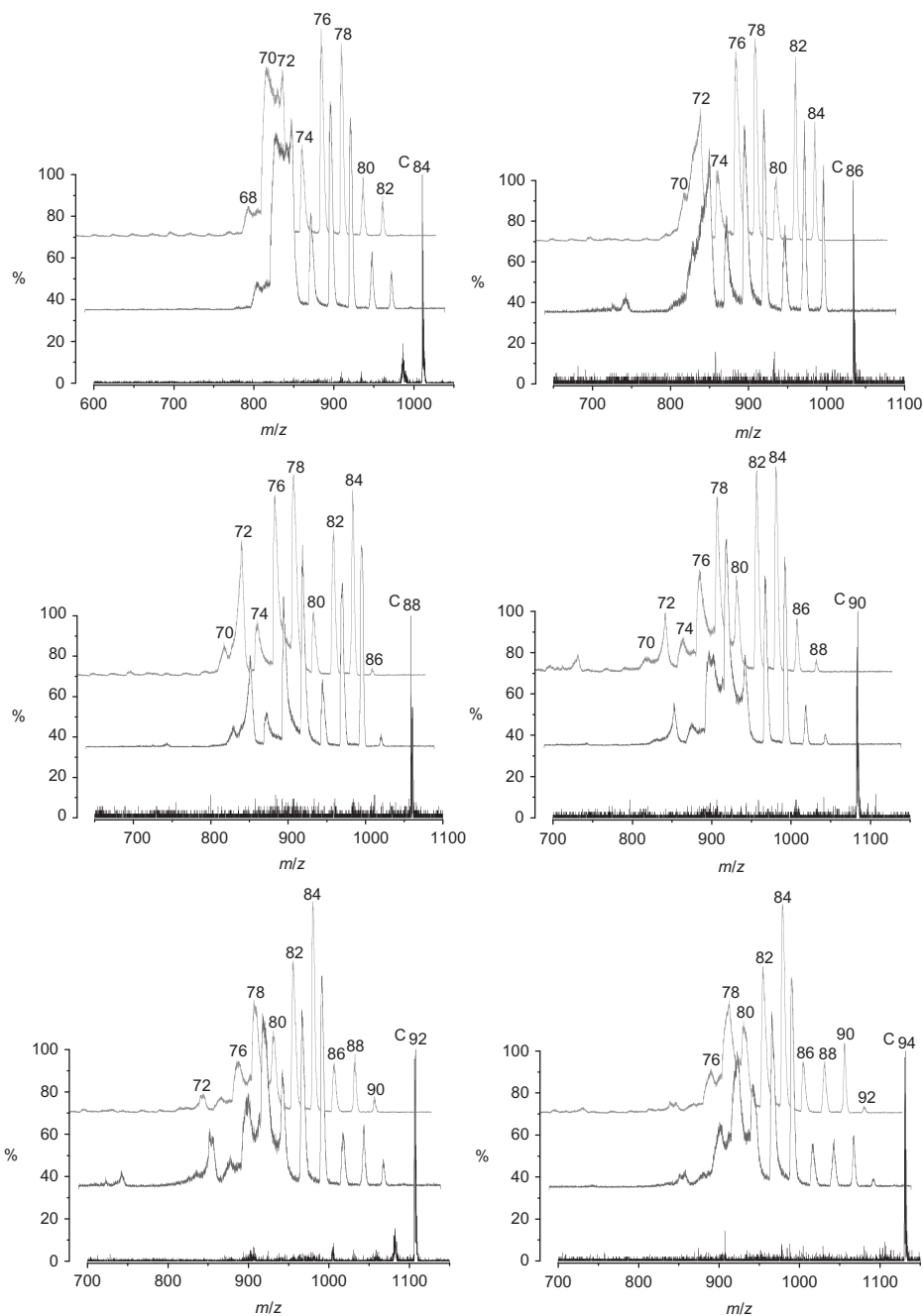


**Figure 1.** Positive-ion LDI-TOF mass spectrum of a commercial mixture of fullerenes.



**Figure 2.** Positive-ion LDI-TOF/TOF mass spectra of  $C_{60}$ ,  $C_{70}$ ,  $C_{76}$ ,  $C_{78}$ ,  $C_{80}$ , and  $C_{82}$ . For each fullerene: threshold laser power (bottom), high laser power (middle), and high laser power and CID gas (Ar) switched on (top).

All data are summarized in Table 1. All fullerenes decompose by loss of even numbers of carbon atoms (2, 4, 6, . . . 16). No low-mass  $C_n^+$  fragments were observed in our product ion spectra.



**Figure 3.** Positive-ion LDI-TOF/TOF mass spectra of C<sub>84</sub>, C<sub>86</sub>, C<sub>88</sub>, C<sub>90</sub>, C<sub>92</sub>, and C<sub>94</sub>. For each fullerene: threshold laser power (bottom), high laser power (middle), high laser power and CID gas (Ar) switched on (top). The effect of introducing Ar (under computer control, increasing the pressure to approximately  $6 \times 10^{-6}$  mbar) to act as a CID gas was minimal in these experiments, there was slightly more fragmentation for C<sub>60</sub><sup>+</sup>, but for the higher fullerenes the effect was barely discernible.

**Table 1**  
Peak intensities of fragment ions in the LDI-TOF/TOF of  $C_n^+$ .\*

$C_n^+$	Fullerene selected for MS/MS											
	60	70	76	78	80	82	84	86	88	90	92	94
48	4							3	0–4 %			
50	7							8	5–9 %			
52	9							19	10–19 %			
54	19							28	20–29 %			
56	50							33	30–39 %			
58	11	10						56	> 40 %			
60	⊗	70	7					⊗	Selected fullerene			
62		9	17	6								
64		4	3	3								
66		6	8	8	4							
68		2	26	28	19	11	6					
70		⊗	32	39	48	49	38	15	6	4		
72			6	7	6	16	20	20	16	8	4	
74			2	8	8	6	7	9	8	7	4	
76			⊗	2	11	10	13	20	23	19	15	11
78				⊗	4	7	11	15	20	22	24	26
80					⊗	1	3	5	5	10	12	15
82						⊗	2	10	10	15	16	16
84							⊗	6	10	13	17	18
86								⊗	1	3	5	4
88									⊗	1	4	4
90										⊗	1	4
92											⊗	1
94												⊗

\*Under the harshest conditions attainable in the mass spectrometer, that is, 100% laser power, CID gas (Ar) switched on.

$C_{60}^+$  itself seems particularly resistant to fragmentation, the most abundant fragment ion,  $C_{56}^+$ , represents the loss of only four carbon atoms (probably as 2  $C_2$  molecules). This observation is in stark contrast to the behaviour of the higher fullerenes under the same conditions, for which the most abundant daughter ions involve loss of between 6 (for  $C_{76}^+ \rightarrow C_{70}^+$ ) and 16 (for  $C_{94}^+ \rightarrow C_{78}^+$ ) carbon atoms in total. Particular product ions were especially prominent regardless of the precursor.  $C_{70}^+$  readily lost five pairs of carbon atoms to form  $C_{60}^+$ , much in preference to loss of 1, 2, 3, 4, or 6. Similarly,  $C_{70}^+$  itself was the most abundant product ion in the MS/MS spectra of  $C_{76}^+$ ,  $C_{78}^+$ ,  $C_{80}^+$ ,  $C_{82}^+$ , and  $C_{84}^+$ .  $C_{76}^+$  was the most prevalent in when fragmenting  $C_{86}^+$  and  $C_{88}^+$ .  $C_{78}^+$  is also especially prominent, appearing as the most common fragment from  $C_{88}^+$ ,  $C_{90}^+$ ,  $C_{92}^+$ , and  $C_{94}^+$  precursor ions. Notably, these ions —  $C_{60}^+$ ,  $C_{70}^+$ ,  $C_{76}^+$ , and  $C_{78}^+$  — are also those most abundant in the original mass spectrum of the mixture of higher fullerenes (Figure 1). There are also similarities with the high temperature annealing experiment of Cross and Saunders, in which  $C_{76}$  formed  $C_{70}$  and  $C_{60}$ ,  $C_{78A}$  formed  $C_{78B}$ ,  $C_{76}$ ,  $C_{70}$  and  $C_{60}$ , and  $C_{78B}$  formed  $C_{78A}$ ,  $C_{76}$ ,  $C_{70}$  and  $C_{60}$ . Of course, our experiment is over much more quickly and thus captures many of the kinetic products of  $C_2$  loss rather than the thermodynamic products accessed by the high temperature oven and extended exposure (~40 seconds). Hence we see all the same products—and they are generally the most abundant—but we also observe all the intermediate species as well. Kroto came up with a simple set of rules that predicted certain fullerenes should display special stability, including  $C_{60}$  and  $C_{70}$  (12).

Early theoretical studies of  $C_{20}$  through  $C_{100}$  found  $C_{60}$ ,  $C_{70}$  and  $C_{84}$  to be energetically more stable than their neighbors (13), and simple Huckel MO computations favoured  $n = 60, 70, 76, 78, 84,$  and  $90$  (14).

## Experimental

Mass spectrometric data were collected on a Bruker AutoflexII MALDI-TOF/TOF instrument. The fullerene mixture (“fullerite”) was purchased from Aldrich and used without further purification. A saturated  $CS_2$  solution of the fullerenes was prepared and single drops allowed to evaporate on an AnchorChip<sup>TM</sup> target (15). New spots were used for each experiment. MS/MS was performed using the LIFT experiment. The three spectra collected for each  $C_n^{+\bullet}$  were recorded under conditions of (a) threshold laser power, (b) moderate laser power, and (c) high laser power with the introduction of argon gas for CID. These parameters allowed representative coverage of the fragmentation processes. Each LDI mass spectrum represents the accumulation of between 50 and 500 laser shots on the target, the stage was moved over the course of each experiment to expose fresh sample. Data were not processed (subtraction, smoothing, etc.) in any way. Table 1 was generated by integrating peak areas for each fullerene fragment ion and expressing the intensity as a percentage of the total ion current. This method is likely to slightly overestimate the intensity of the less abundant ions due to contributions from noise.

## Conclusions

LDI-TOF/TOF mass spectrometry has been found to be a simple yet informative technique for probing the fragmentation pathways of higher fullerenes, without the necessity for tedious separations of the individual fullerene components. This methodology should be applicable for the investigation of other polyhedral inorganic clusters.

## Acknowledgments

JSM thanks the Royal Society of Chemistry for a Journals Grant for International Authors. WH thanks the University of Waikato for materials and instrumentation.

## References

1. Cross, R. J. and Saunders, M. (2005) *J. Am. Chem. Soc.*, 127: 3044–3047.
2. (a) Dyson, P. J., Hearley, A. K., Johnson, B. F. G., Langridge-Smith, P. R. R., and McIndoe, J. S. (2004) *Inorg. Chem.*, 43: 4962–4973; (b) McIndoe, J. S. (2003) *Trans. Met. Chem.*, 28: 122–131; (c) Hearley, A. K., Johnson, B. F. G., McIndoe, J. S., and Tuck, D. G. (2002) *Inorg. Chim. Acta*, 334: 105–112; (d) Johnson, B. F. G. and McIndoe, J. S. (2000) *Coord. Chem. Rev.*, 200–202: 901–932.
3. (a) McElvany, S. W., Ross, M. M., and Callahan, J. H. (1992) *Acc. Chem. Res.*, 25: 162–168; (b) McElvany, S. W. and Ross, M. M. (1992) *J. Am. Soc. Mass Spectrom.*, 3: 268–280.
4. Kroto, H. W., Heath, J. R., O’Brien, S. C., Curl, R. F., and Smalley, R. E. (1985) *Nature*, 318: 162–163.
5. O’Brien, S. C., Heath, J. R., Curl, R. F., and Smalley, R. E. (1988) *J. Chem. Phys.*, 88: 220–230.
6. Doyle, R. J. and Ross, M. M. (1991) *J. Phys. Chem.*, 95: 4954–4956.
7. Beck, R. D., Rockenberger, J., Weis, P., and Kappes, M. M. (1996) *J. Chem. Phys.*, 104: 3638–3650.

8. Cordero, M. M., Cornish, T. J., and Cotter, R. J. (1996) *J. Am. Soc. Mass Spectrom.*, 7: 590–597.
9. Diederich, F. and Whetten, R. L. (1992) *Acc. Chem. Res.*, 25: 199–126.
10. Suckau, D., Resemann, A., Schüerenberg, M., Hufnagel, P., Franzen, J., and Holle, A. (2003) *Anal. Bioanal. Chem.*, 376: 952–965.
11. Kurogochi, M., Matsushita, T., and Nishimura, S. I. (2004) *Angew. Chem. Int. Ed.*, 43: 4071–4075.
12. Kroto, H. W. (1987) *Nature*, 329: 529–531.
13. Wang, C. Z., Zhang, B. L., Ho, K. M., and Wang, X. Q. (1993) *Int. J. Mod. Phys.*, 7: 4305–4329.
14. Liu, X., Schmalz, T. G., and Klein, D. J. (1992) *Chem. Phys. Lett.*, 188: 550–554.
15. Schüerenberg, M. and Franzen, J. U.S. patent 6,287,872, 2001.