PS #6 Practice Questions – Mechanisms (Mannich Reaction; Electrophilic Aromatic Substitution), Chemoselective Reduction of Carbonyl Groups

Part I. In the Mannich reaction a secondary amine (R_2NH) is condensed with formaldehyde to form a reactive imine salt. The imine salt is reacted with any enolizable aldehyde or ketone to form an amino-ketone product (Mannich base). Show the detailed mechanism for the formation of the Mannich base (**Step 1**), then alkylate and eliminate to form the enone (**Step 2**). Refer to pages 712 – 715 in Clayden and Greeves.

 $\textbf{Part II.} \ \text{Show the detailed mechanism for the electrophilic aromatic substitution of 2,6-dimethylanisole with Br_2.}$

Part III. The following reduction reactions will not work as written. State why and propose an alternate set of conditions that will carry out the desired transformation.

1.
$$\bigcup_{0}^{0}$$
 \bigcup_{0}^{0} \bigcup_{1}^{0} \bigcup

4.
$$C_5H_7$$

$$C_5H_7$$

$$C_5H_7$$