

# Utility of a multi-tracer approach as a component of adaptive monitoring for municipal wastewater impacts

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## ABSTRACT

Distinguishing municipal wastewater effluent (MWW) from other industrial effluents or through an urbanized watershed can be challenging. In complex receiving environments, linking environmental responses to specific compounds or effluents is not always straight forward. In order to characterize the inherent complexity of tracing MWW in aquatic systems influenced by multiple stressors, a proposed multi-tracer suite is intended to highlight areas of potential biological concern. Characterization and quantification of effluent exposure to aquatic biota in this manner is essential to shape policies intended to encourage wastewater infrastructure development (i.e. treatment plant upgrade) and broader environmental management. This paper describes the use of a comprehensive suite of tracers that includes isotopes in support of a core surveillance program, demonstrating its effectiveness both empirically and with respect to diagnostic value contributed to monitoring programs.

**Key words** | emerging contaminants, environmental tracers, monitoring design, wastewater

## HIGHLIGHTS

- Characterizing presence of trace organic compounds.
- Developing predictive capacity of effects using water chemistry.
- Linking effects-driven water chemistry analysis (science) to decision making (policy).

## INTRODUCTION

In Canada, there are 3,700 municipal wastewater treatment plants that release 6 trillion liters of treated municipal wastewater effluent (MWW) annually, presenting a major threat

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to aquatic environments (Chambers *et al.* 1997); over 20% of this release occurs into Alberta waters (Government of Canada 2010). Growing populations are continually

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introducing emerging contaminants comprising mainly of organic pollutants into the waste streams through broad consumption and disposal. This phenomenon has increased the need to trace the fate of effluent in the environment. It is therefore the intent of this paper to identify the value in using multiple tracers to guide where and when intervention should occur in waters receiving MWWE and when upgrades to infrastructure may be required. Ideal tracers for MWWE support characterization of exposure in the receiving environment and thus should be wastewater-specific and not ubiquitous in the environment (Gasser *et al.* 2014). Additional properties of good tracers as presented by Lalor & Pitt (1999) are:

1. Significant difference in pollutant concentrations between possible source waters;
2. Small variations in pollutant concentrations within each likely source water;
3. Conservative behavior (i.e. concentrations do not change due to physical, chemical or biological processes); and
4. Ease of measurement with adequate detection limits, good sensitivity and repeatability.

As these properties are specific to quantification and resolution, capability of tracers to produce meaningful data can be evaluated by considering desirable properties of indicators (Cairns *et al.* 1993):

- Cost-effective
- Historical data
- Lack of redundancy
- Timeliness
- Biologically relevant
- Socially relevant
- Interpretable
- Anticipatory
- Integrative
- Diagnostic

This subset of applicable properties of indicators (those overlapping with tracer properties and not relevant to water quality were omitted) presented in the seminal work by Cairns *et al.* (1993) ensures that compounds being quantified in the environment are both relevant and meaningful when combined with tracer objectives. Current MWWE discharge regulations are limited and place wastewater effluent quality

limits on biological oxygen demand, suspended solids, residual chlorine and unionized ammonia (Government of Canada 2012). These analytes, unfortunately, do not provide insight into fate and potential responses of receiving environments due to MWWE exposure and specifically to the over 40,000 trace organic compounds currently manufactured, consumed and potentially present in MWWE (Diamond *et al.* 2011). Biological organisms exposed to MWWE accumulate compounds which may lead to impacts (Wang *et al.* 2011). It is possible to characterize exposure and identify areas where potential impacts would be most likely using a multi-tracer method.

A multi-tracer approach for MWWE leverages the utility of multiple analytes to enable the modeling of fate, degradation, potential synergistic or additive effects and food-web dynamics. This study proposes the use of trace organic compounds in concert with isotopes and metals to maximize diagnostic potential of receiving environment water quality monitoring for MWWE. The suite of analytes needs to incorporate a variety of chemicals that have differing characteristics, the potential to 'pass-through' wastewater treatment systems, and are relatively easy to measure. The utility of anti-epileptics such as carbamazepine as well as nonsteroidal anti-inflammatory drugs such as naproxen and ibuprofen in tracing wastewater exposure in receiving environments has been well documented due to the tendency of these chemicals to be recalcitrant through treatment (Metcalf *et al.* 2003; Conkle *et al.* 2008). Illicit drugs (Metcalf *et al.* 2010), artificial sweeteners (Liu *et al.* 2014) and caffeine (Thomas & Foster 2005) have also been shown to have utility as tracers for MWWE. Metals, although more ubiquitous in the environment are introduced in significant quantities through MWWE and in dissolved form can elicit toxic responses (Gagnon & Saulnier 2003; AEP 2018). Isotopes are incorporated into the multi-tracer analytical suite to enable confirmation of exposure of resident aquatic organisms to wastewater and to trace exposure through food-webs (Loomer *et al.* 2015; Robinson *et al.* 2016). In addition to determining the spatial extent of biotic exposure, isotopes can also enable discrimination between multiple complex effluents introduced to rivers in more developed watersheds (Dubé *et al.* 2005).

To test the potential utility of the approach in a system with a relatively high level of dilution, samples of river

water were collected in the Bow River in Calgary, Alberta and analyzed for several potential tracers to track wastewater from a major municipal wastewater treatment plant (MWWTP). MMWE is a significant point-source stressor on the Bow River, with the Bonnybrook MWWTP releasing the largest volume of treated effluent of the three MWWTPs within the City of Calgary. The Bonnybrook MWWTP is a tertiary treatment plant with an average discharge per day in 2011 and 2012 of 300,000 m<sup>3</sup>/day released on the left bank of the river (V Moore, pers. comm. 2019). Treatment technologies include ammonia, nitrogen, phosphorous control and ultra-violet disinfection (Alberta Environment 2009). The objective of the study was to assess the persistence of several target analytes relative to conventional wastewater performance parameters to determine their effectiveness as tracers and to define their diagnostic value.

## MATERIALS AND METHODS

### Water sampling

Samples of river water were collected at 10 locations (Figure 1) over a distance of 14.8 km along the Bow River through the city of Calgary (southern Alberta, Canada) in October 2011 and May 2012. Samples were obtained at half-depth from a flat-bottomed boat along the channel thalweg, except at the sewage outfall location BR-2 (this sample collected from the middle of the plume). Winter samples could not be obtained due to safety and access limitations. The locations are labeled BR-1 through BR-10. BR-1 is upstream of the Bonnybrook WWTP. Average discharge of the Bow River was measured by combining flow along the main stem, with the Elbow River and Nose Creek flows (sum of Water Survey Canada stations 05BH004, 05BJ001 and 05BH901, respectively, with diverted flow at 05BM015 subtracted) and was 94.6 m<sup>3</sup>/s for sampling in October of 2011 and 93.6 m<sup>3</sup>/s for sampling in May 2012. These conditions are representative of 'low flow' conditions in the Bow River within Calgary, where the Bonnybrook MWWTP effluent makes up to 4.5% of the flow in the river (Supplementary Material, Table S1).

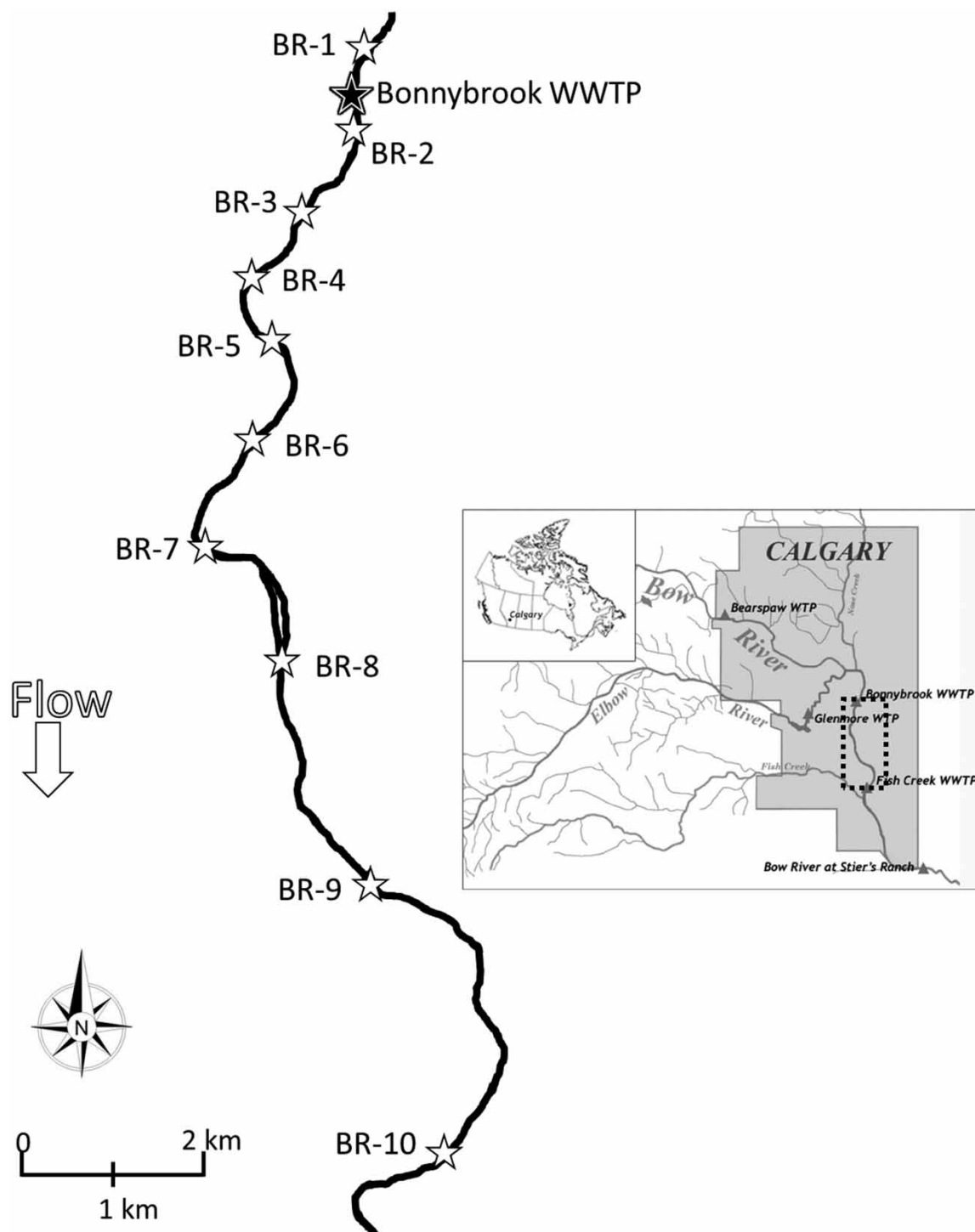
Water samples were collected from the middle of the wastewater plume discharging from the Bonnybrook

WWTP. The wastewater plume was previously delineated in detail by Vandenberg *et al.* (2005). In addition, the plume is clearly visible from publicly available aerial photographs. Measurements of pH, Eh, temperature, electrical conductivity (EC) and dissolved oxygen (DO) were made in the field using a Hydrolab. All probes were calibrated prior to sampling; Eh electrode was calibrated in Zobell's (Nordstrom, 1977) and Light's solutions (Light, 1972); the pH electrode was calibrated with standard buffer solutions at pH 4, 7 and 10. All river samples were collected in a consistent manner utilizing a peristaltic pump and a 6.4 mm (1/4-inch) diameter polyethylene tubing attached to a stainless steel weight and lowered into the upper 0.5 m of the water column. At each location, water samples were collected into pre-cleaned Nalgene bottles for major anions and cations, dissolved metals, dissolved organic carbon (DOC), nutrients (NH<sub>3</sub>-N, NO<sub>2</sub>-N, NO<sub>3</sub>-N, PO<sub>4</sub>-P), H<sub>2</sub>S, Fe(II) and stable isotopes  $\delta^{18}\text{O}$ ,  $\delta^2\text{H}$ ,  $\delta^{13}\text{C}_{\text{DIC}}$ ,  $\delta^{13}\text{C}_{\text{DOC}}$ ,  $\delta^{34}\text{S} + \delta^{18}\text{O}_{\text{SO}_4}$ ,  $\delta^{15}\text{N} + \delta^{18}\text{O}_{\text{NO}_3}$ ,  $^{87/86}\text{Sr}$ , and  $\delta^{37}\text{Cl}$ . Dissolved metals and major cations were acidified in the field with 12 N trace-metal grade HNO<sub>3</sub> to a pH of <1 and anions, organic acids, and DOC were unacidified. Samples for trace organic compounds including pharmaceuticals and artificial sweetener were collected in pre-cleaned 1 L amber glass bottle and preserved in the field with trace-grade H<sub>2</sub>SO<sub>4</sub>. All water samples were field filtered with 0.45  $\mu\text{m}$  high-volume inline filters with the exception of  $\delta^{13}\text{C}_{\text{DIC}}$ . All water samples were immediately refrigerated at 4 °C until analysis.

### Water analysis

Water samples were analyzed using inductively coupled plasma optical emission spectrometry (ICP-OES) for major cations, inductively coupled plasma mass spectrometry (ICP-MS) for dissolved metals and ion chromatography (IC) for anions. Combustion and infrared detection was used for DOC analysis. Dissolved H<sub>2</sub>S, Fe(II) and NH<sub>3</sub>-N were determined in-field using a Hach DR 2800 portable spectrometer following the procedure outlined by SMEWW (2005).

The oxygen isotopic composition of water samples was determined by the H<sub>2</sub>O-CO<sub>2</sub> equilibration method. Hydrogen (deuterium) isotopic composition in water was



**Figure 1** | Map of the Bow River within the City of Calgary, Alberta, Canada. 'Bonnybrook WWTP' and the adjacent filled star indicates the effluent outfall of the Bonnybrook municipal wastewater treatment plant. Dotted line within inset map identifies study area as presented. Remaining open stars denote sampling locations visited in the fall of 2011 and spring of 2012 with numbering corresponding to successive locations downstream and data (Figures 2–6).

measured on an isotope ratio mass spectrometer (MS) determined by reduction of water to H<sub>2</sub> gas using chromium metal as an active reducing agent. All isotopic ratios are reported as ‰ against an internationally accepted standard. The results of oxygen and deuterium isotope measurements are expressed in delta notation ( $\delta^{18}\text{O}$  and  $\delta^2\text{H}$ ) relative to Vienna Standard Mean Ocean Water (V-SMOW) standard. The  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$  composition of groundwater was measured to identify the degree of evaporative enrichment. The method of Assayag *et al.* (2006) was used to determine  $\delta^{13}\text{C}$  in dissolved inorganic carbon (DIC), reported relative to Vienna Pee Dee Belemnite (V-PDB) standard. The  $\delta^{13}\text{C}$  of DIC was used to identify processes such as carbonate dissolution and bacterial reduction. Values of  $\delta^{13}\text{C}$ -DOC were determined using continuous flow isotope ratio mass spectrometry after preparation to remove the inorganic carbon using phosphoric acid and saturated potassium persulfate. The  $\delta^{13}\text{C}$ -DOC results were reported relative to V-PDB with an uncertainty of  $\pm 0.3\text{‰}$ . Sulfur-isotope ratios of dissolved sulfate were determined by adding BaCl<sub>2</sub> to precipitate BaSO<sub>4</sub>, which was then converted into SO<sub>2</sub> in an elemental analyzer coupled to a MS in continuous flow mode (Giesemann *et al.* 1994). Sulfur-isotope measurements are reported in delta notation ( $\delta^{34}\text{S}$ ) relative to the Vienna Canyon Diablo Troilite (V-CDT) standard. Oxygen-isotope analysis on BaSO<sub>4</sub> samples was carried out by pyrolysis in a high temperature conversion/elemental analyzer – interfaced to an isotope-ratio MS. Boron isotope abundance measurements were made on water samples that were purified by microsublimation (Grobe *et al.* 2000) and analyzed on a thermal ionization MS (Williams *et al.* 2001). Strontium was extracted from the waters using ion exchange resin columns, eluted with deionized water, evaporated and then analyzed using positive ion thermal ionization mass spectrometry with a Thermo Electron Triton (PTIMS) instrument using the method described by Burton *et al.* (2002) using reference material SRM987. Measurement uncertainty in the  $^{87}\text{Sr}/^{86}\text{Sr}$  ranged was  $\pm 0.0001$ . Chlorine isotope analysis was measured by continuous flow isotope ratio mass spectrometry using the method described by Shouakar-Stash *et al.* (2005). The method involved the precipitation of dissolved chloride to silver chloride (AgCl), followed by the conversion of silver chloride to methyl chloride (CH<sub>3</sub>Cl). Methyl chloride was then isolated on a GC column and the

isotope ratio was measured using an IRMS. Results are reported relative to Standard Mean Ocean Chloride (SMOC) with an uncertainty of  $\pm 0.2\text{‰}$ .

ACE-K and the eight pharmaceutical compounds were analyzed using solid-phase extraction (SPE) and high-performance liquid chromatography (HPLC) followed by tandem mass spectrometry using previously published methods (Vanderford & Snyder 2006; Stafiej *et al.* 2007; Scheurer *et al.* 2009) with slight modifications such as inclusion of isotope dilution techniques for each target compound and optimization of instrument operating conditions. Native compounds were supplied by Sigma-Aldrich (Oakville, Canada) with the exception of ACE-K obtained from Toronto Research Chemicals Inc. (Toronto, Canada). Isotope labeled standards, including ACE-K-d4 and SMX-d4 (Toronto Research Chemicals Inc., Toronto, Canada), MDA-d5 and MDMA-d5 (Cerilliant Inc., Texas, USA), and CBZ-d10, CAF-d3, IBU-d3, GEM-d6, and [13C]-NAP (Cambridge Isotope Laboratory Inc., Cambridge, USA) were obtained as dry powders. Standard stock solutions of 1,000 mg L<sup>-1</sup> were prepared by dissolving each compound in methanol. Working standard solutions containing all analytes were prepared by serial dilution in methanol: water 50:50 vol/vol. HPLC-grade ammonium acetate and formic acid were obtained from Sigma-Aldrich (Oakville, Canada). Ultrapure (Type 1) water was generated using a Milli-Q A10 water system (18.2 MU cm @ 25 °C).

Prior to analysis, 600 mL aliquots of aqueous samples were spiked with a consistent amount of internal standard mixture. These samples were passed through SPE cartridges (Oasis HLB 6 mL glass cartridges; Waters Corp., Mississauga, Canada) pre-conditioned with 3 × 2 mL methanol and then equilibrated with 3 × 2 mL ultrapure water. After loading the 600 mL samples, the cartridges were washed using 3 × 2 mL 5 vol% methanol then eluted with 3 × 2 mL methanol. The eluate was collected in an amber glass vial and stored at 4 °C until analysis.

The extracts were analyzed using an Agilent 1100 HPLC (Agilent Technologies, Mississauga, Canada) followed by electrospray tandem mass spectrometry (MS/MS; 4000 Q TRAP, Applied Biosystems, Foster City, USA). Caffeine, SMX, CBZ, MDA and MDMA were analyzed in electrospray ionization positive (ESI+) mode, while IBU, GEM, NAP and CE-K were analyzed in ESI negative (ESI<sup>-</sup>)

mode. The gradient and mobile phases were changed from previous methods based on the HPLC analytical columns and MS requirements. The compound and source-dependent parameters of the MS were modified to obtain an optimum signal response. The mobile phases for analysis in ESI<sup>+</sup> mode consisted of 0.1% formic acid and 5 mM ammonium acetate in water (phase A) and 99.9% MeOH with 0.1% formic acid (phase B). A gradient elution started at 10% B for 3 min, increased to 90% B in 10 min and then held at 90% B min for 10 min. The flow rate was 1,000 mL min<sup>-1</sup> and the injection volume was 15 mL. The mobile phases for analysis in ESI<sup>-</sup> mode consisted of 6.9 mM acetic acid in 300 mL acetonitrile and 700 mL water (phase A) and 100% acetonitrile (phase B). A gradient elution started at 12% B, then increased linearly to 40% in 10 min and held at 40% for 10 min. The injection volume was 10 mL and the flow rate was 1,000 mL min<sup>-1</sup>. For ACE-K analyzed in ESI<sup>-</sup> mode, the mobile phases consisted of 20 mM ammonium acetate in water (phase A) and 20 mM ammonium acetate in methanol (phase B). The gradient elution started at 2% B, then increased linearly to 75% in 8 min and held at 75% for 8 min. The injection volume was 10 mL and the flow rate was 1,000 mL min<sup>-1</sup>.

## Statistics

All data were subjected to quality assurance and quality control measures outlined in Surface Water Quality Programs in Alberta (Alberta Environment 2006), with values below detection limit set to 0. Principal component analysis (PCA) was used to determine the reduced number of latent variables (principal components) explaining the most variance in the dataset. Samples were normalized using the STANDARDIZE function in Microsoft Excel (version 1908) to a mean of 0 and a variance of 1 prior to analysis. This undirected method served to visually represent patterns of a large number of variables which are spatially autocorrelated. Variables following a correlation with traditional wastewater indicators such as chloride are expected to coalesce with larger component loadings across component 1. The PCA was completed with IBM SPSS Statistics 26 using Varimax Rotation to explain the influence of various tracers. Analytes which were significantly correlated to chloride, a conventional tracer of wastewater, were explored graphically using normalized data.

Correlational analyses were conducted using IBM SPSS Statistics 26 using a Pearson Correlation test with a two-tailed test of significance ( $\alpha$ ) at 0.01 with raw nonnormalized data.

## RESULTS AND DISCUSSION

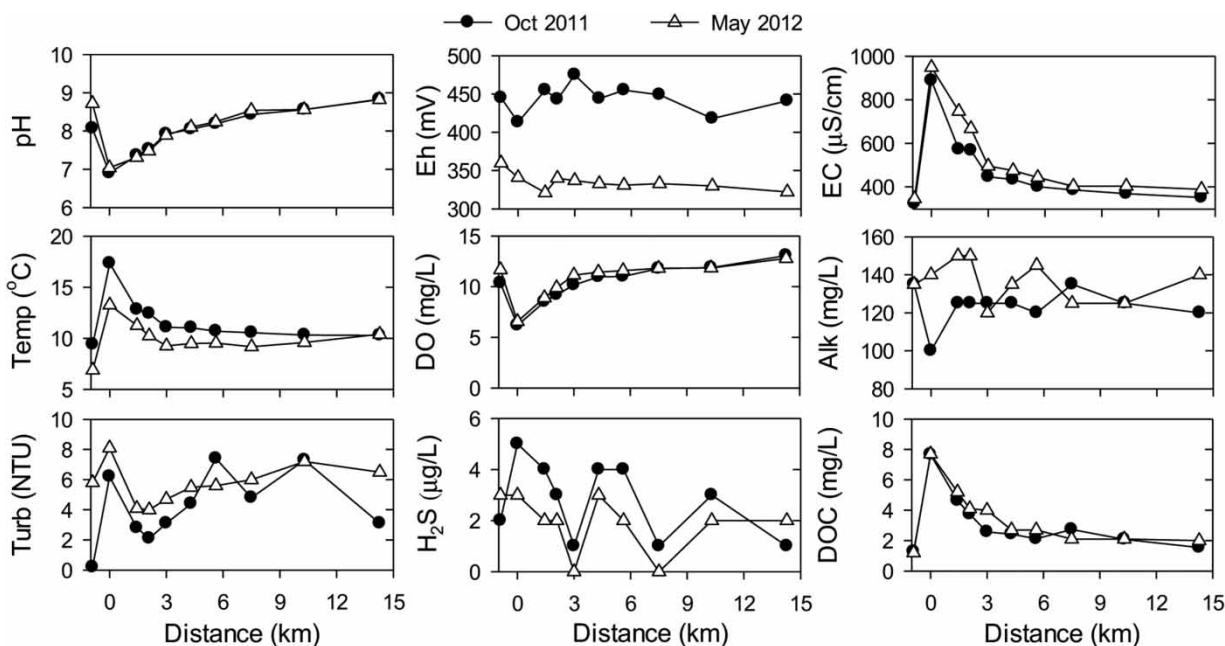
### Conventional parameters, major ions and nutrients

Flow conditions in both sampling campaigns were similar (94.6 and 93.6 m<sup>3</sup>/s for 2011 and 2012, respectively) and thus supported comparisons in both summer (May) and fall (October) conditions. DO levels consistently averaged ~10 mg/L in both years (Figure 2), with the lowest concentrations in the river immediately downstream of the effluent plume due to increased oxygen demand. Similarly, pH values were consistent between years with average pH at ~8, with the lowest values (~7–7.3) observed at locations immediately downstream of the discharge point as is characteristic of MWWE. Reduction potential showed minimum variation across sites within years but showed a higher reduction potential in fall of 2011, which is believed to be the driver in differences in ammonia concentrations (Figure 3) between years. Temperature differentials between effluent and receiving environment were observed as warmer conditions (up to 7 °C) caused by effluent release. All major ions and nutrients (Figure 3) were considerably elevated and attenuated across downstream sampling locations, corresponding to increased conductivity as expected. With exposure profile for traditional (albeit ubiquitous) ions such as chloride established, these compounds provided the basis to conduct statistical comparisons to correlate and assess exposure against. Conventional parameters commonly have guidelines associated with them due to the increased potential for acute toxicity; however, the other classes of compounds in this study are more likely to result in chronic toxicity.

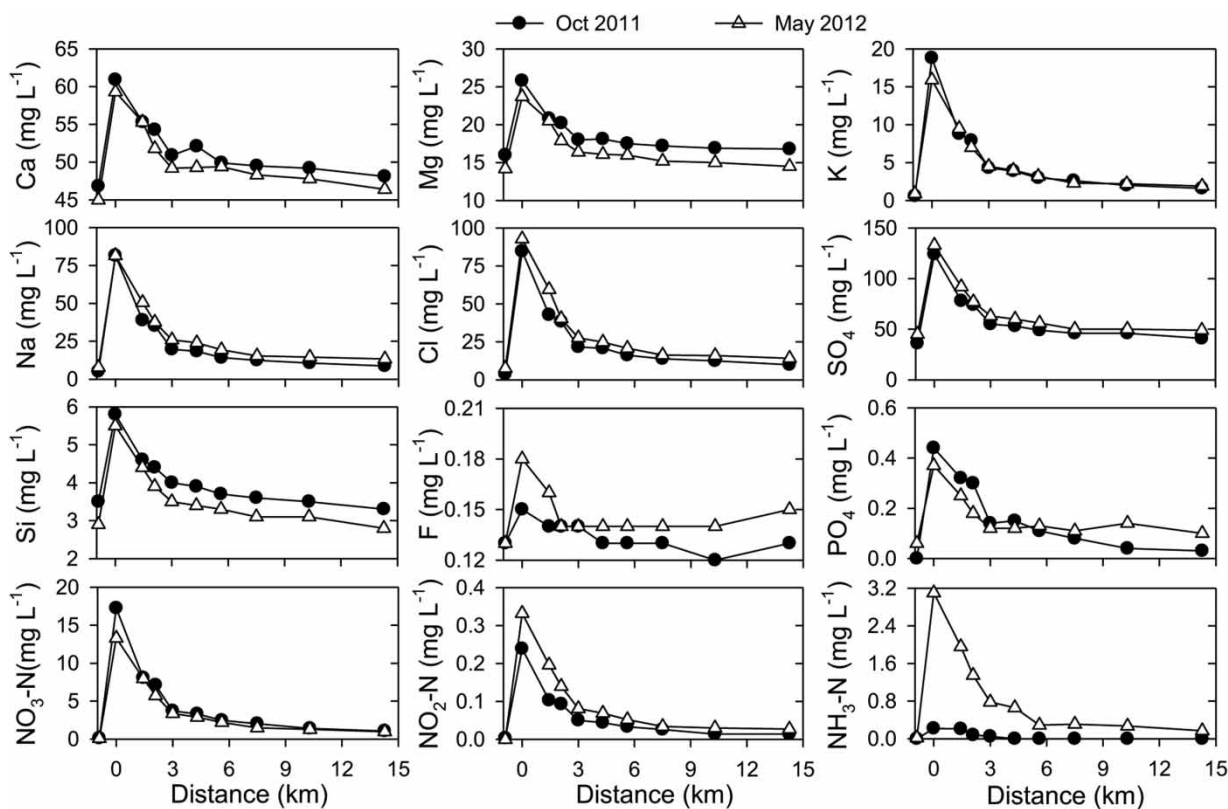
### Metals

Dissolved metals showed consistent patterns of more than doubling at the discharge point relative to upstream concentrations (Figure 4) with the exception of dissolved barium which was diluted by wastewater effluent. Dissolved iron was substantially lower downstream of the discharge point

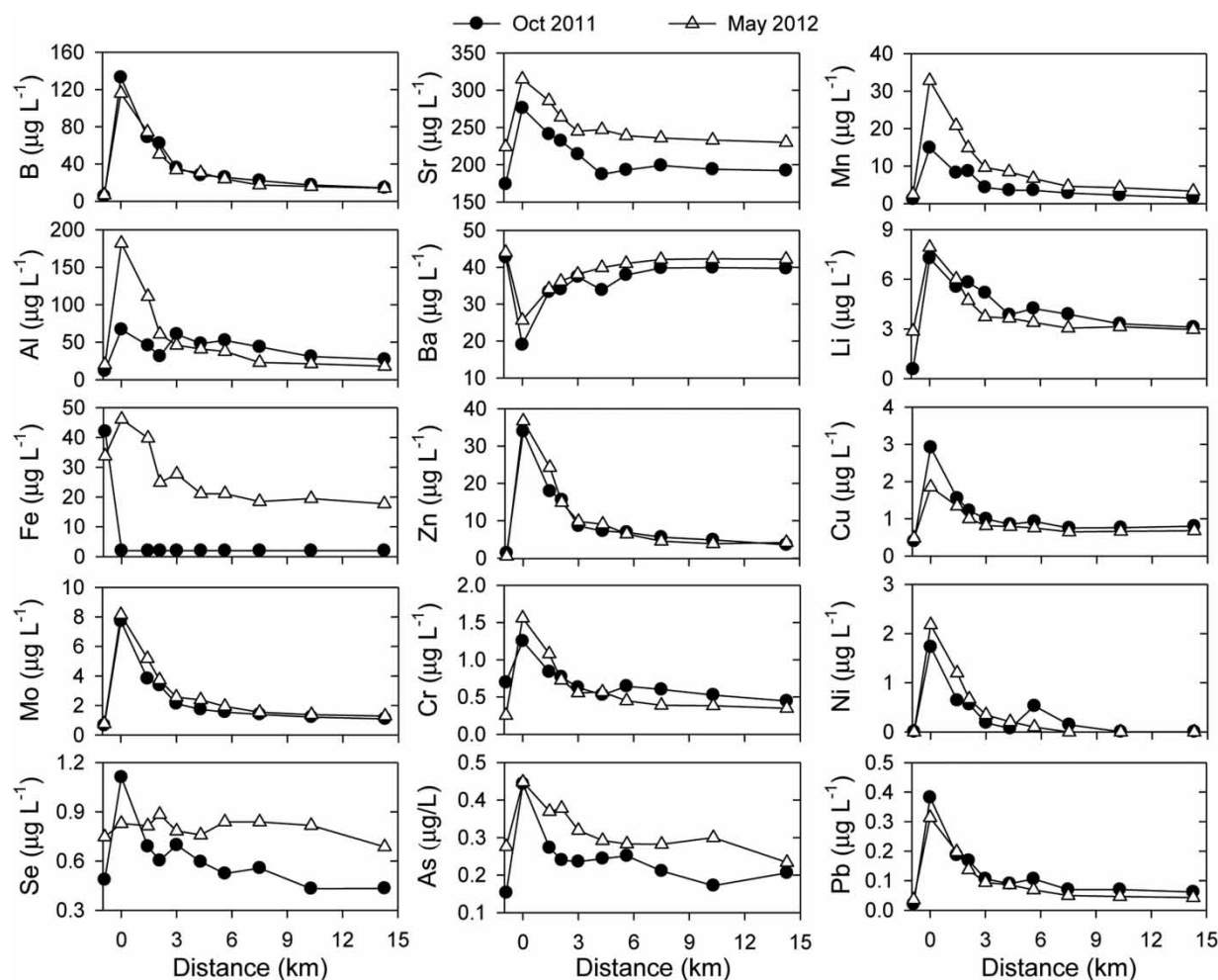




**Figure 2** | Field parameters and conventional wastewater performance analyte concentrations in samples obtained from the Bow River across a MWWE gradient.



**Figure 3** | Major ion and nutrient parameters including nitrogen sulfate and phosphate expressed as concentration in samples obtained from the Bow River across a MWWE gradient.



**Figure 4** | Dissolved metal concentrations in samples obtained from the Bow River across a MWWE gradient.

in fall of 2011, which may have been a result of higher oxidation-reduction potential across sites in the fall despite similar flows. Increases in dissolved metals at the outfall may have been further exacerbated by low pH which can act to bring metals which are present into solution. These spikes in metal concentrations were sustained and took up to 15 km to attenuate to background levels. Metals are well documented as presenting a potential for toxicity to invertebrates and chronic toxicity to fish via direct toxicity (e.g. gill damage) or at critical life stages (i.e. embryo), resulting in environmental protection guidelines or 'limits' (AEP 2018). These guidelines do not necessarily account for site specific sensitivities and are continuously being re-evaluated based on toxicity mechanisms (AEP 2018). This knowledge combined with the quantities of metals released via

MWWE makes metals relevant and nonredundant in the tracer suite from both a concentration and a total loading perspective.

### Trace organic compounds

Trace organic compounds including pharmaceuticals quantified in this study are compounds which are known to be recalcitrant through treatment processes and are commonly used to characterize effluent exposure. Anti-epileptic drugs like carbamazepine and artificial sweeteners like acesulfame are largely resistant to biodegradation (Metcalf *et al.* 2003; Liu *et al.* 2014) making them good candidate tracers for MWWE. Despite high dilution capacity afforded by the receiving environment, artificial sweeteners, anti-epileptic,



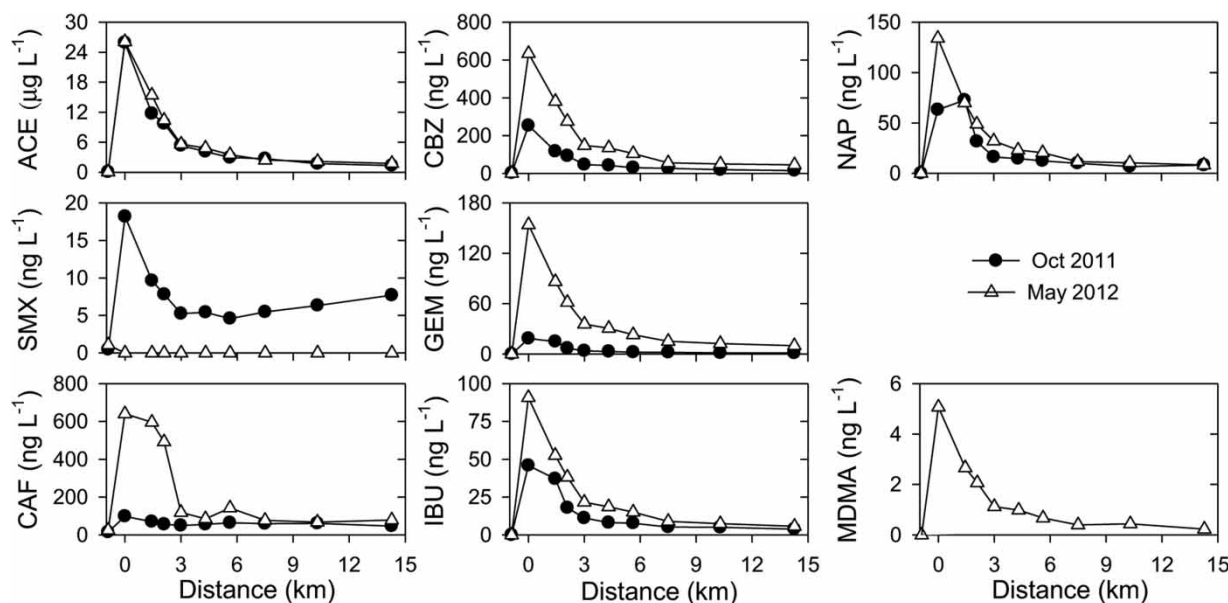
anti-inflammatory and lipid controlling compounds all showed varying effluent-driven increases downstream of the discharge point (Figure 5). The reducing environment observed through summer conditions and oxidizing in the fall appear to impact concentration and fate of pharmaceuticals in a manner similar to iron and ammonia.

### Stable isotope ratios

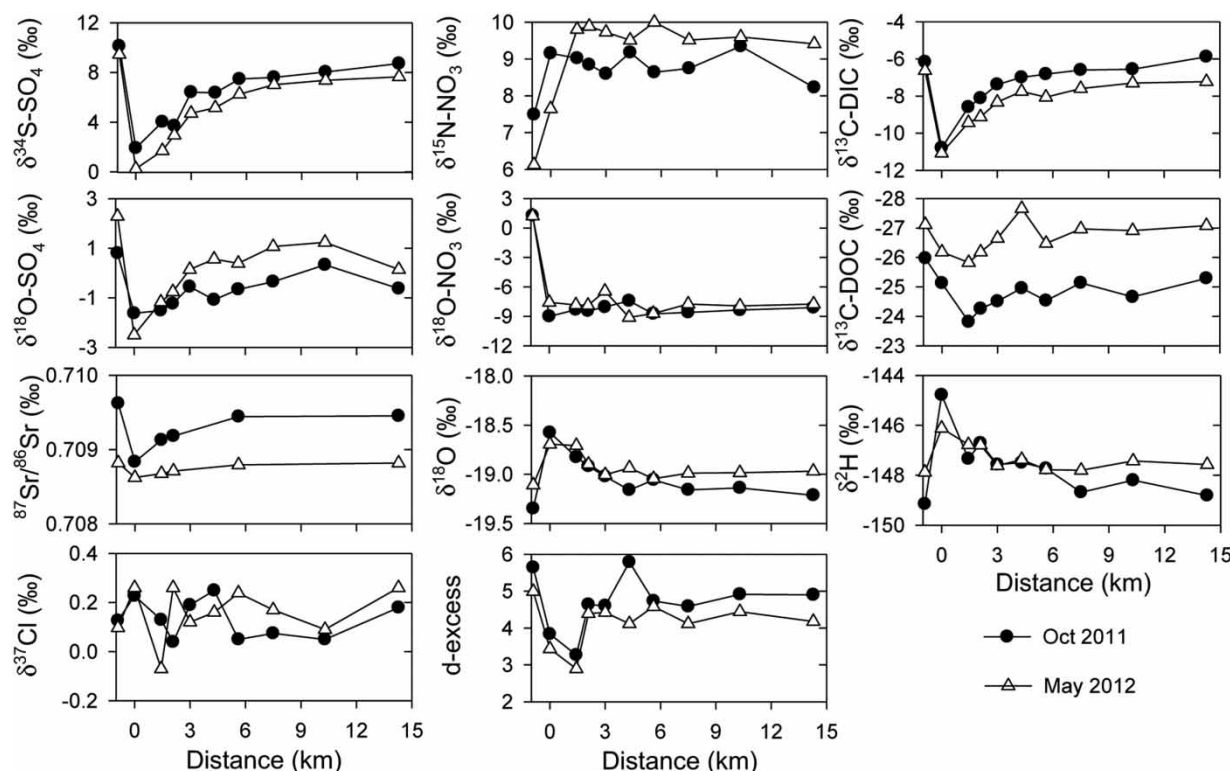
In analysis of the stable isotope compositions of water: sulfate, nitrate, DIC, DOC, strontium and chloride showed patterns, which corresponded in varying degrees to effluent discharge (Figure 6)  $\delta^2\text{H}$  and  $\delta^{18}\text{O}$  increases in receiving water at the effluent outfall can be attributed to evaporation and 'off-gassing' of  $\text{O}_2$  and  $\text{H}_2$  through the treatment process (Wayland & Hobson 2001; Bedard-Haughn *et al.* 2003). Conversely, decreases in  $\delta^{13}\text{C}$  of DIC and to a lesser extent  $\delta^{13}\text{C}$  of DOC in the outfall relative to upstream and downstream conditions are a product of the large quantities of terrestrially derived organic material entering treatment plants, as well as the microbial treatment process itself (Wayland & Hobson, 2001; Bridgeman *et al.* 2014).  $\delta^{34}\text{S}$  and  $\delta^{18}\text{O}$  values decreased from +10 and +2‰ above the effluent to +1 and −2‰, respectively downstream of the effluent,

due to the low  $\delta^{34}\text{S}$  and  $\delta^{18}\text{O}$  values of effluent sulfate (Kruk *et al.* 2020). Downstream of the effluent  $\delta^{34}\text{S}$  and  $\delta^{18}\text{O}$  values of sulfate increased to +8 and +1‰ due to mixing with riverine sulfate.  $\delta^{15}\text{N}$  values of nitrate increased from +7 ‰ above the effluent to +10 ‰ downstream of the effluent input, while  $\delta^{18}\text{O}$  value of nitrate decreased markedly from +2‰ to −9‰ due to the influx of effluent nitrate with elevated  $\delta^{15}\text{N}$  and very low  $\delta^{18}\text{O}$  values (Kruk *et al.* 2020). As sampling was conducted during characteristic low flow periods where a higher level of groundwater influence is observed, Sr minimal differences in isotope ratios are presumed to be driven by groundwater recharge (Armstrong *et al.* 1998). No evidence of effluent impact was apparent in Cl isotope ratios.

Isotopes are important to verifying that changes appearing in potential biological treatment can be correctly attributed to wastewater. Bi-plots between isotope ratios for specific compounds can show distinct signatures or 'fingerprints' that can provide greater utility in verification of exposure in biological receptors and food web dynamics to better inform cumulative effects assessment at the watershed scale. Similar patterns in isotope ratios as seen in a number of compounds have been observed previously in wastewater studies and specifically with carbon isotopes; however, these



**Figure 5** | Dissolved concentrations of acesulfame (ACE), select pharmaceuticals (CBZ – Carbamazepine; GEM – Gemfibrozil; IBU – Ibuprofen; NAP – Naproxen; SMX – Sulfamethoxazole), caffeine (CAF) and illicit drug Ecstasy (MDMA) in samples obtained from the Bow River across a MWW effluent gradient.



**Figure 6** | Stable isotope concentration ratios measured in water obtained from the Bow River across a MWWE gradient.

patterns and directionality of relative differences are also reliant upon the nature of the receiving environment (Kuuppo *et al.* 2006; Leavitt *et al.* 2006; Loomer *et al.* 2015).

### Statistical analysis

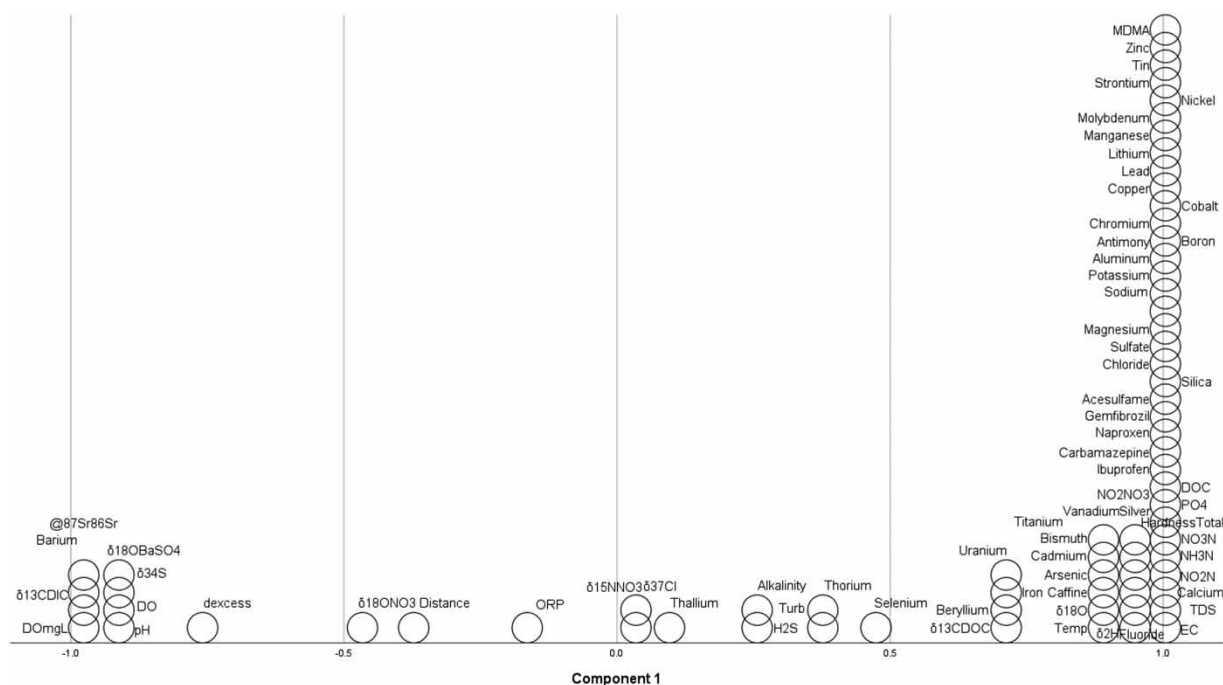
Results from a PCA resulted in 56.9% component loadings on Principal Component 1 and 8.2% on Principal Component 2 due to the high correlation between analytes in response to wastewater release. In comparing these compounds through PCA, compounds which had strong correlations with traditional wastewater indicators such as chloride showed high positive loadings on Component 1 (Figure 7). Conversely where values for tracers followed a negative correlation against wastewater as observed with pH, these were represented as high negative loadings on Component 1. Isotope ratios such as  $\delta^{18}\text{O-NO}_3$  and  $\delta^{15}\text{N-NO}_3$  did not correspond to such patterns and in turn remained closer to the origin on Component 1 and instead showed greater loadings on Component 2 with

modifying factors such as oxidation-reduction potential, alkalinity and distance from effluent release.

Pearson product moment correlation coefficients were calculated for tracers against chloride to identify correlations with two-tailed significance at  $\alpha = 0.01$ . Prior to evaluating data for correlations against chloride, correlation against field EC was first confirmed ( $r = 0.991$ ,  $p = 5.1 \times 10^{-10}$ ). Insignificant correlations were limited to those between chloride and: turbidity, ORP, alkalinity, selenium, thallium, thorium, uranium, sulfamethoxazole,  $\delta^{13}\text{CDOC}$ ,  $\delta^{15}\text{NNO}_3$ ,  $\delta^{18}\text{ONO}_3$  and  $\delta^{37}\text{Cl}$ . Figures representing significant correlations ( $p \leq 0.01$ ,  $r > 0.8$  or  $r < -0.8$ ) graphically for conventional parameters, major ions and nutrients, metals, trace organics and isotopes are included as Supplementary material, Figures S1 through S5.

### Utility of a multi-tracer approach

The appeal in utilizing multiple tracers in monitoring wastewater effluent is (A) the added specificity in detecting the wastewater signal in the receiving environment and (B) the



**Figure 7** | Plot of correlation values from component matrix for all analytes obtained from the Bow River across a MWE gradient. Principal components extraction resulted in 56.9% of loadings on Component 1 and 8.2% loadings on Component 2 (details provided in Supplementary material, Table S2).

development of a chemical baseline to interpret potential effects-based (biological) changes. Rivers that flow through urban centers are often influenced by a number of sources of environmental substances of concern. These sources can include but are not limited to agriculture, urban runoff, hospital effluents, septic tanks, subsurface storage of household and industrial waste, landfill leachate and others. When faced with the need to distinguish the signal of wastewater and characterize its potential influence in a system to manage wastewater infrastructure: conventional parameters, metals, isotopes and pharmaceuticals and personal care products collectively meet the criteria of good tracers and desirable properties of indicators to support these needs.

Establishing a multi-tracer suite of compounds for MWE that incorporates traditional compounds (i.e. metals), wastewater performance indicators (i.e. chloride and ammonia), pharmaceuticals and isotopes in concert enables a dynamic baseline to be established. Benefits of this approach in the context of desirable properties of the indicators and tracers presented in this study are summarized in Table 1. These select compounds can be utilized to characterize local receiving environments and enable comparisons

within and across watersheds impacted by wastewater effluent release. In the study completed by Liu *et al.* (2014) on the Grand River, similar concentrations of pharmaceuticals and other analytes were observed across sampling locations despite higher and lower dilution ratios relative to the current study (Supplementary Material Table S1). It is important to note that through the period of time in which the survey on the Grand River was completed, that parallel research programs indicated biochemical, organ and community level responses in fish in response to effluent release (Tetreault *et al.* 2011, 2013; Tanna *et al.* 2013). Although the pharmaceutical tracers utilized in both the present study and by Liu *et al.* (2014) are not directly implicated in biological responses, they serve as general tracers of organic compounds to establish site conditions where a risk to aquatic wildlife might exist. In studies where impacts to wildlife are observed, isotopic analysis has been critical in confirming that biological effects are elicited by wastewater exposure (Dubé *et al.* 2005).

Characterization of receiving environments using a multi-tracer approach with simultaneous biological monitoring can help to establish adverse outcome pathways (Ankley

**Table 1** | Summary of tracer categories and associated tracers for use as a comprehensive suite in routine monitoring of wastewater influenced systems, with biological relevance and relevant criteria for selection identified. (Tracer criteria 1 through 4 presented in Introduction)

Tracer category	Proposed tracers	Fate	Diagnostic capacity	Biological relevance	Tracer criteria met	Indicator criteria met
Conventional wastewater parameters	pH, temperature, conductivity, turbidity, DO, dissolved organic carbon	N/A – physicochemical in nature	Low – confounded by other point and nonpoint sources	Indicative of potential acute toxicity	1, 2, 4	Cost-effective Historical data Timeliness Socially relevant Interpretable
Major ions and nutrients	Chloride, nitrate, nitrite, phosphate	Remain in solution and are processed or consumed	Low – confounded by runoff and other point and nonpoint sources	Drive primary productivity and acute toxicity	1, 2, 4	Biologically relevant Socially relevant Broadly applicable
Metals	Aluminum, arsenic, barium, boron, copper, chromium, iron, manganese, molybdenum, nickel, lead, lithium, selenium, strontium, zinc*	Largely sequestered in treatment plant sludge, dissolved metals remain in solution, influenced by pH	Medium – Known driver of toxicity impacting development and respiration	Known driver of toxicity impacting development and respiration	1, 2, 3, 4	Biologically relevant Diagnostic
Trace organic contaminants	Carbamazepine, gemfibrozil, fluoxetine, sulfamethoxazole, naproxen, ibuprofen	Sequestered in sludge, in solution, adsorb to sediment based on $K_{ow}$	High – Specific to wastewater inputs (including septic systems) with no overlap with other point-sources/non-point sources	Known driver of toxicity via endocrine disruption	1, 2, 3, 4	Biologically relevant Lack of redundancy Anticipatory Diagnostic
Isotopes	C-dic/doc, N-nitrate, o-nitrate, s-sulfate, o-sulfate, N, C, O18, H2	N/A – Signatures influenced by mixture of effluent influenced by plant and animal metabolism and evaporation	Neutral – Provides baseline signature that includes an understanding of effluent mixture and contributions of other source waters	Confirms direct and indirect exposure to invertebrates and higher order organisms	1, 2, 4	Biologically relevant Interpretable Integrative

\*Metals are analyzed as a suite of 'regulated metals' by laboratories corresponding to freshwater aquatic life guidelines (AEP 2018).

*et al.* 2010) that are MWWE-specific and support targeted investigation of specific chemical compounds should impacts be observed. This approach enables the ability to establish conditions where MWWE poses a risk to receptors and supports development of predictive capacity related to effluent quantity and quality. Upon establishing baseline conditions, where there is an absence of degradation in biological data or trends in chemistry, sampling intensity in time and space can be adapted (Arciszewski *et al.* 2017) to reflect these conditions.

Of pollutants that are released through MWWE, trace organic compounds can be of the highest potency but are the most varied and can elicit a response below analytical detection limits. For this reason, effects are not easily generalized across sites in the context of specific or groups of compounds within effluent as complex as MWWE. The proposed multi-tracer suite used in conjunction with established biological criteria for MWWE (Kilgour *et al.* 2005) can serve to support stronger comparisons and better identify scenarios that merit further investigation of trace organic compounds and potential management.

Consistent collection of data using a multi-tracer suite enables important questions to be asked of the data. Specifically, we can adapt the questions identified by Diamond *et al.* (2010) for screening of high-risk CECs and adapt them to tracers for surveillance monitoring by asking:

- Where tracers (some or all) are elevated, do we find biological effects?
- Are tracers elevated where we find effects?
- Where tracers are found to be elevated, what type of conditions (i.e. physical or chemical) are present that create risk?

Further validation of tracers is necessary through synoptic study across multiple MWWE releases in effluent dominated systems. Specifically, planned next steps include sampling across two additional downstream outfalls, two tributaries and a series of stormwater outfalls within the watershed, leveraging historical data collected by the City of Calgary. Identifying areas of exposure and potential concern in a multi-tracer approach enables the characterization of relevant chronic effects where they may exist.

This approach is thus seen as adaptive due to forward compatibility considering ever-changing production and

consumption of high risk (i.e. organic) compounds entering MWWTPs. Ultimately, receiving environments which are not monitored for change via surveillance monitoring which considers the fate of stressors leaves wastewater infrastructure operators and environmental managers unable to plan and adapt appropriately. Implementation of a multi-tracer approach enables exploration of current conditions in the context of historical conditions and to consider the relevance of US and European chemistry frameworks that consider pollutants introduced by wastewater for provincial and federal regulation revisions. Due to the absence of effects-directed monitoring in existing federal wastewater regulations, local efforts to ensure ecosystem health and in-turn continued protection of ecosystem services, in particular to meet the objectives of Alberta's renewed Water for Life strategy (Government of Alberta 2008).

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## SUPPLEMENTARY MATERIAL

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## REFERENCES

- Alberta Environment 2006 *Guidelines for Quality Assurance and Quality Control in Surface Water Quality Programs in Alberta*. Prepared by Patricia Mitchell Environmental Consulting, Project Number W0603.
- Alberta Environment 2009 *Municipal Wastewater Facility Assessment Phase 2 Database Application*. Prepared by AECOM Canada Ltd, Project Number 106396.



- Alberta Environment and Parks (AEP) 2018 *Environmental Quality Guidelines for Alberta Surface Waters*. Water Policy Branch, Policy Division, Edmonton, Alberta, Canada.
- Ankley, G. T., Bennett, R. S., Erickson, R. J., Hoff, D. J., Hornung, M. W., Johnson, R. D., Mount, D. R., Nichols, J. W., Russom, C. L., Schmieder, P. K. & Serrano, J. A. 2010 *Adverse outcome pathways: a conceptual framework to support ecotoxicology research and risk assessment*. *Environmental Toxicology and Chemistry: An International Journal* **29** (3), 730–741.
- Arciszewski, T. J., Munkittrick, K. R., Scrimgeour, G. J., Dubé, M. G., Wrona, F. J. & Hazewinkel, R. R. 2017 *Using adaptive processes and adverse outcome pathways to develop meaningful, robust, and actionable environmental monitoring programs*. *Integrated Environmental Assessment and Management* **13** (5), 877–891.
- Armstrong, S. C., Sturchio, N. C. & Hendry, M. J. 1998 *Strontium isotopic evidence on the chemical evolution of pore waters in the Milk River Aquifer, Alberta, Canada*. *Applied Geochemistry* **13** (4), 463–475.
- Assayag, N., Rivé, K., Ader, M., Jézéquel, D. & Agrinier, P. 2006 *Improved method for isotopic quantitative analysis of dissolved inorganic carbon in natural water samples*. *Rapid Communications in Mass Spectrometry* **20**, 2243–2251.
- Bedard-Haughn, A., Van Groenigen, J. W. & Van Kessel, C. 2003 *Tracing  $^{15}\text{N}$  through landscapes: potential uses and precautions*. *Journal of Hydrology* **272** (1–4), 175–190.
- Bridgeman, J., Gulliver, P., Roe, J. & Baker, A. 2014 *Carbon isotopic characterisation of dissolved organic matter during water treatment*. *Water Research* **48**, 119–125.
- Burton, G. R., Morgan, V. I., Boutron, C. F. & Rosman, K. J. F. 2002 *High-sensitivity measurements of strontium isotopes in polar ice*. *Analytical Chimica Acta* **469** (2), 225–233.
- Cairns, J., McCormick, P. V. & Niederlehner, B. R. 1993 *A proposed framework for developing indicators of ecosystem health*. *Hydrobiologia* **263** (1), 1–44.
- Chambers, P. A., Allard, M., Walker, S. L., Marsalek, J., Lawrence, J., Servos, M., Busnarda, J., Munger, K. S., Adare, K., Jefferson, C. & Kent, R. A. 1997 *Impacts of municipal wastewater effluents on Canadian waters: a review*. *Water Quality Research Journal* **32** (4), 659–714.
- Conkle, J. L., White, J. R. & Metcalfe, C. D. 2008 *Reduction of pharmaceutically active compounds by a lagoon wetland wastewater treatment system in Southeast Louisiana*. *Chemosphere* **73** (11), 1741–1748.
- Diamond, J., Thornton, K., Bartell, S., Munkittrick, K., Kidd, K., Kapo, K., Latimer, H., Gilliam, J. & Gerritsen, J. 2010 *Development of Diagnostic Tools for Trace Organic Compounds and Multiple Stressors*. Water Environment Research Foundation.
- Diamond, J., Thornton, K., Munkittrick, K., Kidd, K., Bartell, S. & Kapo, K. 2011 *Diagnostic Tools to Evaluate Impacts of Trace Organic Compounds*. Water Environment Research Foundation, Alexandria, VA, USA.
- Dubé, M. G., Benoy, G. A., Blenkinsopp, S., Ferone, J. M., Brua, R. B. & Wassenaar, L. I. 2005 *Application of multi-stable isotope ( $^{13}\text{C}$ ,  $^{15}\text{N}$ ,  $^{34}\text{S}$ ,  $^{37}\text{Cl}$ ) assays to assess spatial separation of fish (longnose sucker *Catostomus catostomus*) in an area receiving complex effluents*. *Water Quality Research Journal* **40** (3), 275–287.
- Gagnon, C. & Saulnier, I. 2003 *Distribution and fate of metals in the dispersion plume of a major municipal effluent*. *Environmental Pollution* **124** (1), 47–55.
- Gasser, G., Pankratov, I., Elhanany, S., Glazman, H. & Lev, O. 2014 *Calculation of wastewater effluent leakage to pristine water sources by the weighted average of multiple tracer approach*. *Water Resources Research* **50** (5), 4269–4282.
- Giesemann, A., Jäger, H. J., Norman, A.-L., Krouse, H. R. & Brand, W. A. 1994 *On-line sulfur-isotope determination using an elemental analyzer coupled to a mass spectrometer*. *Analytical Chemistry* **66**, 2816–2819.
- Government of Alberta 2008 *Water for Life Strategy – A Renewal*. Government of Alberta, Edmonton, Alberta. Available from: <https://www.alberta.ca/water-for-life-strategy.aspx> [accessed 11 October 2018].
- Government of Canada 2010 *Canada Gazette. Government Notices: Wastewater Systems Effluent Regulations*, Vol. 144 (12). Government of Canada, Ottawa, Ontario. Available from: <http://www.gazette.gc.ca> [accessed 29 December 2018].
- Government of Canada 2012 *Canada Gazette. Fisheries Act: Wastewater Systems Effluent Regulations*, Vol. 146 (15). Government of Canada, Ottawa, Ontario. Available from: <http://www.gazette.gc.ca> [accessed 29 December 2018].
- Grobe, M., Machel, H. G. & Heuser, H. 2000 *Origin and evolution of saline groundwater in the Munsterland Cretaceous Basin, Germany: oxygen, hydrogen, and strontium isotope evidence*. *Journal of Geochemical Exploration* **69** (70), 5–9.
- Kilgour, B. W., Munkittrick, K. R., Portt, C. B., Hedley, K., Culp, J., Dixit, S. & Pastershank, G. 2005 *Biological criteria for municipal wastewater effluent monitoring programs*. *Water Quality Research Journal* **40** (3), 374–387.
- Kruk, M. K., Mayer, B., Nightingale, M. & Laceby, J. P. 2020 *Tracing nitrate sources with a combined isotope approach ( $\delta^{15}\text{NNO}_3$ ,  $\delta^{18}\text{ONO}_3$  and  $\delta^{11}\text{B}$ ) in a large mixed-use watershed in southern Alberta, Canada*. *Science of the Total Environment* **703**, 135043.
- Kuuppo, P., Tamminen, T., Voss, M. & Schulte, U. 2006 *Nitrogenous discharges to the eastern Gulf of Finland, the Baltic Sea: elemental flows, stable isotope signatures, and their estuarine modification*. *Journal of Marine Systems* **63** (3–4), 191–208.
- Lalor, M. & Pitt, R. 1999 *Use of tracers to identify sources of contamination in dry weather flow*. *Watershed Protection Techniques* **3** (1), 12.
- Leavitt, P. R., Brock, C. S., Ebel, C. & Patoine, A. 2006 *Landscape-scale effects of urban nitrogen on a chain of freshwater lakes*



- in central North America. *Limnology and Oceanography* **51** (5), 2262–2277.
- Light, T. S. 1972 Standard solution for redox potential measurements. *Analytical Chemistry* **44** (6), 1038–1039.
- Liu, Y., Blowes, D. W., Groza, L., Sabourin, M. J. & Ptacek, C. J. 2014 Acesulfame-K and pharmaceuticals as co-tracers of municipal wastewater in a receiving river. *Environmental Science: Processes & Impacts* **16** (12), 2789–2795.
- Loomer, H. A., Oakes, K. D., Schiff, S. L., Taylor, W. D. & Servos, M. R. 2015 Use of stable isotopes to trace municipal wastewater effluents into food webs within a highly developed river system. *River Research and Applications* **31** (9), 1093–1100.
- Metcalfe, C. D., Miao, X. S., Koenig, B. G. & Struger, J. 2003 Distribution of acidic and neutral drugs in surface waters near sewage treatment plants in the lower Great Lakes, Canada. *Environmental Toxicology and Chemistry: An International Journal* **22** (12), 2881–2889.
- Metcalfe, C., Tindale, K., Li, H., Rodayan, A. & Yargeau, V. 2010 Illicit drugs in Canadian municipal wastewater and estimates of community drug use. *Environmental Pollution* **158** (10), 3179–3185.
- Nordstrom, D. K. 1977 Thermochemical redox equilibria of ZoBell's solution. *Geochimica et Cosmochimica Acta* **41** (12), 1835–1841.
- Robinson, C. S., Tetreault, G. R., McMaster, M. E. & Servos, M. R. 2016 Impacts of a tertiary treated municipal wastewater effluent on the carbon and nitrogen stable isotope signatures of two darter species (*Etheostoma blennioides* and *E. caeruleum*) in a small receiving environment. *Ecological Indicators* **60**, 594–602.
- Scheurer, M., Brauch, H. J. & Lange, F. T. 2009 Analysis and occurrence of seven artificial sweeteners in German waste water and surface water and in soil aquifer treatment (SAT). *Analytical and Bioanalytical Chemistry* **394**, (6), 1585–1594.
- Shouakar-Stash, O., Drimmie, R. J. & Frape, S. K. 2005 Determination of inorganic chlorine stable isotopes by continuous flow isotope ratio mass spectrometry. *Rapid Communications in Mass Spectrometry* **19**, 121–127.
- SMEWW 2005 *Standard Methods for the Examination of Water and Wastewater*, 21st edn. American Public Health Association/American Water Works Association/Water Environment Federation, USA.
- Stafiej, A., Pyrzynska, K. & Regan, F. 2007 Determination of anti-inflammatory drugs and estrogens in water by HPLC with UV detection. *Journal of Separation Science* **30** (7), 985–991.
- Tanna, R. N., Tetreault, G. R., Bennett, C. J., Smith, B. M., Bragg, L. M., Oakes, K. D., McMaster, M. E. & Servos, M. R. 2013 Occurrence and degree of intersex (testis-ova) in darters (*Etheostoma* spp.) across an urban gradient in the Grand River, Ontario, Canada. *Environmental Toxicology and Chemistry* **32** (9), 1981–1991.
- Tetreault, G. R., Bennett, C. J., Shires, K., Knight, B., Servos, M. R. & McMaster, M. E. 2011 Intersex and reproductive impairment of wild fish exposed to multiple municipal wastewater discharges. *Aquatic Toxicology* **104** (3–4), 278–290.
- Tetreault, G. R., Brown, C. J., Bennett, C. J., Oakes, K. D., McMaster, M. E. & Servos, M. R. 2013 Fish community responses to multiple municipal wastewater inputs in a watershed. *Integrated Environmental Assessment and Management* **9** (3), 456–468.
- Thomas, P. M. & Foster, G. D. 2005 Tracking acidic pharmaceuticals, caffeine, and triclosan through the wastewater treatment process. *Environmental Toxicology and Chemistry: An International Journal* **24** (1), 25–30.
- Vandenberg, J. A., Ryan, M. C., Nuell, D. D. & Chu, A. 2005 Field evaluation of mixing length and attenuation of nutrients and fecal coliform in a wastewater effluent plume. *Environmental Monitoring and Assessment* **107** (1–3), 45–57.
- Vanderford, B. J. & Snyder, S. A. 2006 Analysis of pharmaceuticals in water by isotope dilution liquid chromatography/tandem mass spectrometry. *Environmental Science & Technology* **40** (23), 7312–7320.
- Wang, S., Oakes, K. D., Bragg, L. M., Pawliszyn, J., Dixon, G. & Servos, M. R. 2011 Validation and use of in vivo solid phase micro-extraction (SPME) for the detection of emerging contaminants in fish. *Chemosphere* **85** (9), 1472–1480.
- Wayland, M. & Hobson, K. A. 2001 Stable carbon, nitrogen, and sulfur isotope ratios in riparian food webs on rivers receiving sewage and pulp-mill effluents. *Canadian Journal of Zoology* **79** (1), 5–15.
- Williams, L. B., Wieser, M. E., Fennell, J., Hutcheon, I. & Hervig, R. L. 2001 Application of boron to the understanding of fluid-rock interactions in a hydrothermally stimulated oil reservoir in the Alberta Basin, Canada. *Geofluids* **1**, 229–240.

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