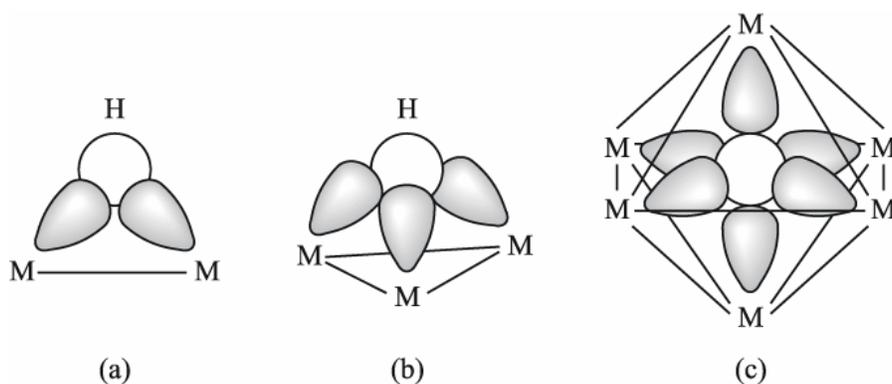
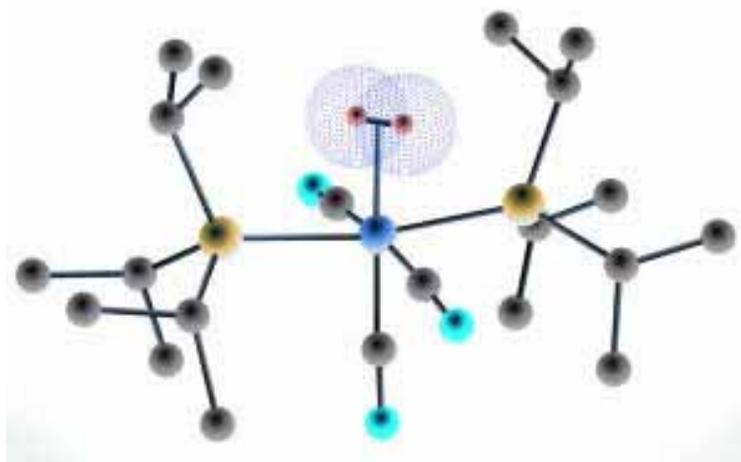


Synthesis of iron carbonyl hydride complexes



Overlap of the H 1s orbital with (a) two or (b) three appropriate metal hybrid orbitals to form $\mu\text{-H}$ and $\mu_3\text{-H}$ bridges. Interstitial hydride ligands (c) are also possible, forming a 7c-2e bond.



The first **dihydrogen** complex: Kubas' $\text{W(CO)}_3(\text{PPr}_3)_2(\eta\text{-H}_2)$

$d_{\text{HH}} = 84 \text{ pm}$ (in free H_2 , 74 pm); $d_{\text{WH}} = 175 \text{ pm}$; $\nu(\text{HH}) = 2690 \text{ cm}^{-1}$
 $\delta_{\text{HH}} = -4.21$ (24 Hz wide), $\delta_{\text{HD}} = -4.21$ (8 Hz wide, 1:1:1 triplet, $J_{\text{HD}} = 33 \text{ Hz}$) cf. $J_{\text{HD}} = 43 \text{ Hz}$ for HD gas
 ^{31}P NMR shows singlet (with W satellites) - J_{HD} is small